



sphere.' As was previously noted, the chemical composition and evolution of the paleoatmosphere were controlled by a strong coupling between the atmosphere, the oceans, the solid earth, and, eventually, the biosphere. The composition was also modified by various atmospheric processes, including photochemical reactions (initiated by the action of solar ultraviolet radiation, which was considerably more intense in the  $O_2$ -deficient paleoatmosphere), chemical reactions, lightning, rainout, and the exospheric escape of light atmospheric gases. Photochemical studies of the paleoatmosphere can be divided into three main areas of research: (1) the photochemistry and stability of the early anaerobic atmosphere, (2) the chemical transition to an oxidizing atmosphere, and (3) the origin and evolution of atmospheric  $O_3$ , with the accompanying shielding of the earth's surface from lethal solar ultraviolet radiation.

Since the early laboratory experiments on chemical evolution, in which complex organic molecules (the precursors of living systems) were synthesized in mixtures of  $NH_3$  and  $CH_4$  exposed to ultraviolet radiation or laboratory electric discharges, it became fashionable to believe that the prebiological primitive atmosphere contained large amounts of  $NH_3$  and  $CH_4$ . However, photochemical calculations indicate that such an early prebiological atmosphere would have been highly unstable against photolysis by solar ultraviolet radiation and, hence, would have been very short lived, on a geological time scale, if it ever existed at all. In addition, there is no geological or geochemical evidence in the rock record to support such a highly reducing early atmosphere. A more mildly reducing atmosphere of  $N_2$ ,  $CO_2$ ,  $CO$ , and  $H_2$  (resulting from volcanic outgassing) is now favored by photochemical, geological, and geochemical considerations. Complex organic molecules have been synthesized in such laboratory mixtures. Recent photochemical calculations also indicate that large amounts of nitrates (formed by lightning) and formaldehyde (formed by atmospheric reactions) could have been transported to the early ocean by precipitation in the  $O_2$ -deficient paleoatmosphere.

The transition from a reducing paleoatmosphere to an oxidizing atmosphere resulted from the build up of atmospheric  $O_2$ . Photochemical calculations indicate that the photolysis of  $H_2O$ , with the accompanying exospheric escape of  $H$ , was probably not a significant source of  $O_2$  over geological time. The inefficient exospheric escape of  $H$  (comparable to today's value) and the volcanic outgassing of  $H_2$  lead to efficient reformation of  $H_2O$ , at the expense of  $O_2$  build up. In the prebiological paleoatmosphere,  $O_2$  was not evenly mixed with altitude, but had a concentration of about  $10^{-12}$  present atmospheric level (P.A.L.) or less at the surface, and a maximum concentration of about  $10^{-6}$  P.A.L. at about 40 km. It appears that photosynthetic activity was the major source of atmospheric  $O_2$ , although there is considerable uncertainty as to the exact chronology for the build up of atmospheric  $O_2$  over geological time.

Recently, a great deal of research has centered on the origin and evolution of  $O_3$ , which was strongly coupled to the build up of  $O_2$  (see Figure 1). The evolution of  $O_3$  and the variation of solar ultraviolet radiation reaching the surface of the earth over geological time (which is controlled by  $O_2$ ) may have had very important implications for the origin and evolution of life on our planet. Studies on the evolution of  $O_3$  have been based on detailed one-dimensional tropospheric/stratospheric photochemical models that include the chemistry of the oxygen, nitrogen, hydrogen, carbon, and chlorine gases. These studies have shed new light on the photochemistry of the paleoatmosphere (for  $O_3$  levels of  $10^{-4}$  P.A.L. to the present). In the  $O_2$ -deficient paleoatmosphere, the enhanced level of solar ultraviolet radiation rendered photolytic destruction the primary loss mechanism for  $N_2O$ ,  $N_2$ , and  $O$  being the products. (In the present atmosphere,  $N_2O$  is largely oxidized to  $NO_x$ , thus becoming involved in the stratospheric ozone cycle.) The enhanced level of solar ultraviolet radiation in the  $O_2$ -deficient paleoatmosphere also resulted in the efficient production of  $OH$  via the photolysis of  $H_2O$  resulting in surface and atmospheric levels of  $OH$  several orders of magnitude greater than in the present atmosphere. The sensitivity of paleoatmospheric  $O_3$  to varying values for solar luminosity, atmospheric temperature, vertical eddy transport, and trace atmospheric gases have been assessed in these studies.

### The Earth's Present Atmosphere

T. E. Graedel (Bell Laboratories) discussed the 'Photochemistry of the Regional Troposphere.' The chemistry of the regional troposphere (a regime defined as encompassing distance scales of the order of 10–1000 km) is intertwined with, but not dominated by, meteorological motions and local emission sources. The time scales of the air motions prescribe the chemical lifetimes of interest, ~1 hour to 5 days, and thus define the species whose chemistry must be studied. These include ammonia, the oxides of nitrogen, the sulfur-containing compounds hydrogen sulfide ( $H_2S$ ) and sulfur dioxide ( $SO_2$ ), and many alkenes, terpenes, aromatics, and aldehydes.

The following are among the regional tropospheric problems of current interest:

1. Downwind Effect. The concentrations of a variety of photochemical products are known to be higher downwind than in the vicinity of the precursor.

## Forum

### Trend Toward Multiple Authors in Research Publications: Failure of the Universities to Support Research

In a previous letter to this column [Fraser-Smith, 1979] one of us drew attention to the marked decline since 1950 in the percentage of single-author papers in the *Journal of Geophysical Research*, *1*, *Space Physics* (JGR 1) and the concomitant increase in the percentage of articles by three or more authors. The decline in single authors is certainly not confined to JGR 1, as is shown by more recent work (according to the Institute for Scientific Information, which indexes 2800 journals, the average number of authors per paper rose from 1.67 to 2.58 between 1960 and 1980) [Broad, 1981], so it seems clear that there is a widespread change taking place in the way scientists report the results of their research. It is perhaps important for us to point out that this is not an academic change; it is taking place right now, and most readers of this column are likely to be affected by it.

There is undoubtedly an element of fashion involved in the decline of single authors, and it may well be that what we are all experiencing individually as scientists is a subtle process of rhinocerization, as described in the play by Ionesco [1960]. However, it would be unusual for a fashion to persist for 30 years or more unless other more substantive factors were involved. The question is, what are these factors?

It would be easy to blame the federal government for the change that is occurring in our reporting habits [see Price, 1981] since funding of research by federal agencies first became significant in the 1950's and, as we all know, it has grown remarkably since. However, an extension of the earlier work on authors, as suggested to us by C. T. Russell, indicates that the changes in the numbers of authors per article are not linked directly to the growth in federal funding. In fact, we will argue that a more important reason for the decline in single authors is a lack of support by our universities.

Following the suggestion by Russell, we reanalyzed our multiple-author data for JGR 1 according to the acknowledged sources of support for the work. Needless to say, our new data are less quantitative than before, but the trends, as illustrated in the figure, are probably accurate.

Shown in the figure are the variations since 1950 in the percentages of papers acknowledging NASA, NSF, or ONR support (these are the three most frequently acknowledged agencies in JGR 1), or no support. Note that it is the percentages of papers in each category that are plotted (one, two, three or more authors), and thus the trends in the number of authors per article previously discussed should not be evident. Clearly, the percentage of papers with no acknowledgement of support has decreased substantially in every author category since 1950, with the most marked decrease occurring in the papers by either one or two authors. Sadly, the percentage of papers acknowledging support from ONR, the first of the U.S. government research funding agencies, has also declined. However, NASA and NSF support has increased substantially and has more than made up for the decline in ONR support. The most interesting feature of the increase since 1980 in the percentage of papers acknowledging support from NASA and NSF is that it takes place in all author categories. In other words, the advent of large-scale federal funding cannot be held directly responsible for the precipitous decline in single-authored papers since 1950.

We might ask why these acknowledgements are so few. It appears that the ready availability of federal funds for research in the past has blinded university research administrations to the advantages of a general research fund, established and increased by gifts and bequests, whose income is used solely to fund internal (or even external) research proposals. Some of the strings normally attached to federal funds, long bemoaned by these administrations, can be avoided, and a wider range of research can be undertaken, in keeping with the particular objectives of the universities. Most important, the disturbing possibility that the federal funds for particular programs of research can be eliminated by the efforts of small groups of congressmen, or even a single U.S. senator, as appears to be happening now in the case of the NASA-funded Search for Extraterrestrial Intelligence (SETI) program, can be more easily avoided.

It might be argued that U.S. universities cannot afford to support research out of their own funds. We disagree. Large sums are raised each year from alumni, companies, and other sources, particularly at what are termed the 'major research universities' (i.e., universities that have been particularly successful at soliciting federal research funds), and even a small diversion of these funds each year from a university research fund (specifically earmarked for research) could soon produce significant income.

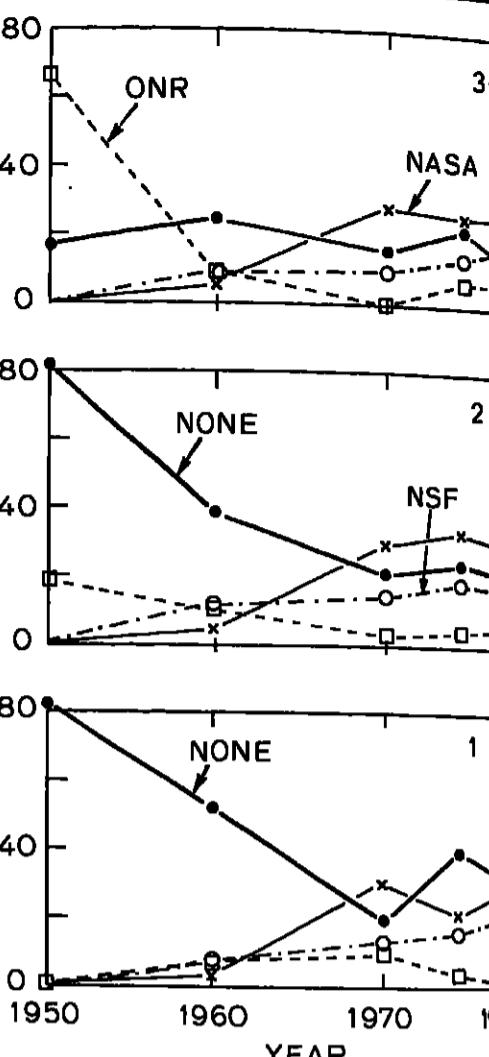


Fig. 1. Variation since 1950 in the acknowledgements of NASA, NSF, and ONR financial support in papers with one (bottom panel), two (middle), and three or more (3+; top panel) authors published in the *Journal of Geophysical Research*. Also shown is the variation in the proportion of papers with no acknowledgement of financial support. The percentages that are plotted apply to each author category.

edged [e.g., Schwind *et al.*, 1960; Rankin and Kurtz, 1970]. We might ask why these acknowledgements are so few. It appears that the ready availability of federal funds for research in the past has blinded university research administrations to the advantages of a general research fund, established and increased by gifts and bequests, whose income is used solely to fund internal (or even external) research proposals. Some of the strings normally attached to federal funds, long bemoaned by these administrations, can be avoided, and a wider range of research can be undertaken, in keeping with the particular objectives of the universities. Most important, the disturbing possibility that the federal funds for particular programs of research can be eliminated by the efforts of small groups of congressmen, or even a single U.S. senator, as appears to be happening now in the case of the NASA-funded Search for Extraterrestrial Intelligence (SETI) program, can be more easily avoided.

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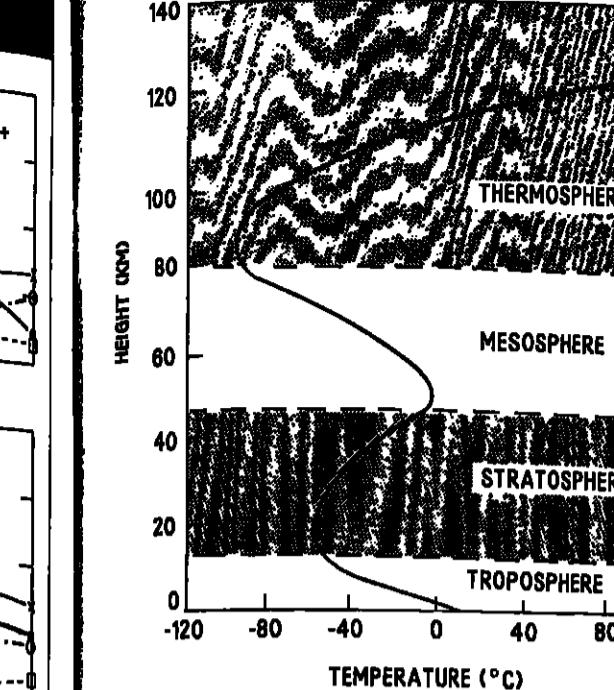


Fig. 2. The four principal layers of the earth's present atmosphere. The boundaries between the layers fluctuate somewhat with time and with geographical location.

meteological motions, with apparently dramatic chemical consequences in some cases. Efforts to model these processes by combining detailed descriptions of organic and inorganic photochemistry and of boundary layer meteorology are in their infancy.

4. Aerosol Nucleation and Growth. Chemical evidence from airborne particulate matter implies that a rich chemistry occurs on particles as a result of their nucleation and growth from reactive gas molecules. Our understanding of these processes is poor and appears limited at present by theoretical uncertainties more than by lack of data, although the latter are surely scanty.

5. Acid Rain. It seems clear that oxides of sulfur and nitrogen emitted in the gas phase are responsible for the increasing acidity of precipitation in northern Europe, northeast United States, and other areas. The rates at which the gas-to-dust transitions occur, and, in fact, the mechanisms of that occurrence, are poorly known. A marriage of atmospheric chemistry, aerosol physics, and cloud physics may be required to reduce the problem to quantitative understanding.

The regional troposphere is probably the most chemically diverse of any of the planetary atmospheric regimes. Perhaps more than any other regime, it must blend emissions, meteorology, and chemistry to analyze properly the processes that occur. Its study is an example of the interdisciplinary requirements of modern atmospheric and planetary science.

William L. Chameides (Georgia Institute of Technology) reviewed the 'Photochemistry of The Global Troposphere.' Key elements of the tropospheric photochemical system are: (1) the production of the free radical  $OH$  in the presence of solar radiation and (2) the emission of reduced gases from the earth's surface. Tropospheric  $OH$  triggers the oxidation of many of the reduced gases generated at the earth's surface and ultimately causes their transformation into chemical forms that are readily removed from the atmosphere by rainout and other heterogeneous processes. This gas-to-particle conversion includes such processes as the nucleation of new particles (after the reaction of  $SO_2$  with  $OH$ , for instance), the condensation of gases onto existing particles (as in the adsorption of  $HNO_3$  onto aerosols), and the heterogeneous reaction of adsorbed gases on particles that serve as catalysts. Aqueous phase chemistry can occur in the liquid layer surrounding solid particles in mists, fog, and clouds.

Aerosols are not only sinks for gases, they can serve as a source when volatile gases are formed on or in the aerosol (as in the release of  $HCl$  from sea-spray) or when cloud droplets evaporate and release their dissolved gases.

We are only just beginning to appreciate the complex role of aerosols in the chemistry of the troposphere. Although we qualitatively understand the processes by which aerosols interact with atmospheric gases, our quantitative understanding is quite poor. Some models for aerosol growth do give fair agreement with chemically simple systems, but current models are generally unable to predict the relative rates of nucleation, condensation, and coagulation of the complex aerosols in the real atmosphere. We need better thermodynamic data for the impure condensed phases that form, as well as much additional theoretical and experimental work on the growth processes. Until we can predict gas-to-particle and particle-to-gas conversion rates, we will be unable to include meaningful source and loss terms in atmospheric trace-gas models. The situation is only slightly better for the measurement of aerosol composition and concentration. Although some non-volatile, noncondensable compounds can be collected and separated nicely, many aerosol measurements are haunted by positive or negative artifacts from the sampling process. Aerosol chemistry desperately needs techniques that can identify specific chemical species.

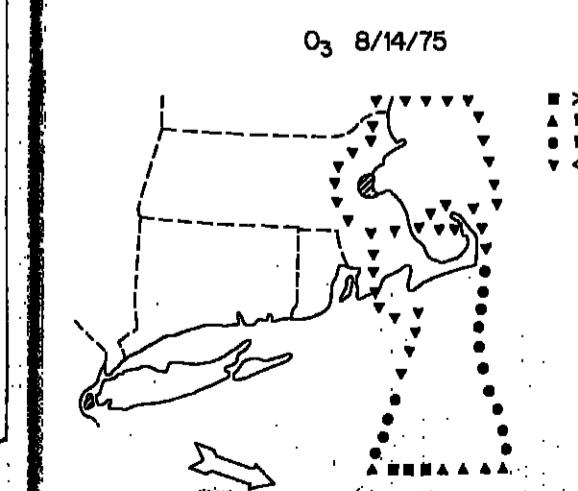


Fig. 3. Ozone concentrations at 390–490 m altitude off the coast of the northeastern United States on the afternoon of August 14, 1975. The highest concentrations were seen about 250 km east of the New York City metropolitan complex, a region of high precursor emission fluxes. Trajectory analysis demonstrated that this high ozone air mass passed over the metropolitan area during the morning of the day on which measurements were made. (After G. W. Siple, C. K. Fitzsimmons, D. F. Zeller, and R. B. Evans, EPA-600/3-77-001a, Enviro. Prot. Agency, Research Triangle Park, N.C., 1977.)

been proposed that they are largely converted to aerosols in the atmosphere, forming the 'blue haze' often seen over forests. Alternatively, it has been proposed that they are chemically fragmented into small molecules, thus constituting a significant global source of  $CO$  and  $H_2$ . The emission fluxes and the fate of the terpenes remain unresolved.

3. Plume Interactions. On the regional scale, emission plumes from cities and from isolated industrial complexes are often brought into contact by

While many uncertainties remain, over the last decade great strides have been made in our understanding of the atmospheric cycles of the carbon compounds  $CH_4$  and  $CO$ , nitrogen oxides, ozone, and halogens. In each of these cycles, it has been found that  $OH$  plays a major role, and thus understanding the processes that control the  $OH$  abundance is of major importance. Because of its reactivity, the concentration of  $OH$  at any location is determined by a balance between photochemical production and destruction.  $OH$  production occurs as a result of the photolysis of  $O_3$  to produce a metastable oxygen atom ( $O(^1D)$ ) followed by the reaction of  $O(^1D)$  with  $H_2O$ . Reactions of  $OH$  with  $CO$  and  $CH_4$  are generally the major loss processes for  $OH$  in the remote troposphere.  $CO$  and  $NO$  can lead to a further enhancement in  $OH$  levels by regenerating  $OH$  from  $HO_2$ , the  $HO_2$  having been produced by the  $CO$  and  $CH_4$  reactions with  $OH$ . Thus, an understanding of  $OH$  concentrations in the atmosphere requires a corresponding understanding of the atmospheric distributions of  $O_3$ ,  $H_2O$ ,  $CO$ ,  $CH_4$ , the nitrogen oxides, and perhaps the array of atmospheric hydrocarbons.

The recent National Science Foundation-sponsored Global Atmospheric Measurements Experiment of Tropospheric Aerosols and Gases (GATE-TAG) included simultaneous measurements of  $OH$  and many of the parameters that influence  $OH$  concentrations:  $CO$ ,  $CH_4$ ,  $O_3$ ,  $H_2O$ , ultraviolet flux, etc. Thus this experiment, by allowing a direct comparison of measured and model-calculated  $OH$ , has made possible the first quantitative test of the photochemical theory of tropospheric  $OH$ . Results for the tropical and subtropical marine boundary layer are quite encouraging. It is suggested, however, that a future project consisting of a more comprehensive set of trace gas measurements would afford a more rigorous test of  $OH$  chemistry and of other fast photochemical processes. It is also noted that vigorous observational programs to elucidate the detailed global distributions of key trace gases and their sources and sinks is still a major need in this field, as is the continued development of sophisticated photochemical models to analyze and simulate these data.

6. Acid Rain. It seems clear that oxides of sulfur and nitrogen emitted in the gas phase are responsible for the increasing acidity of precipitation in northern Europe, northeast United States, and other areas. The rates at which the gas-to-dust transitions occur, and, in fact, the mechanisms of that occurrence, are poorly known. A marriage of atmospheric chemistry, aerosol physics, and cloud physics may be required to reduce the problem to quantitative understanding.

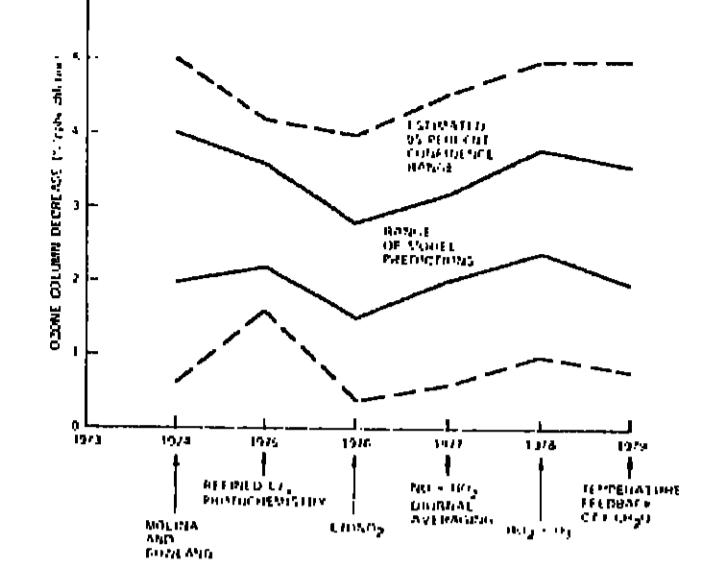
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Ralph J. Cicerone (National Center for Atmospheric Research) summarized the 'Photochemistry of the Stratosphere' (the region of the atmosphere between 10 and 50 km above the surface). The stratosphere's chemistry is controlled by the fluxes of gases from the troposphere and the mesosphere and by their interaction with solar ultraviolet and visible radiation. A major concern of scientists is the photochemistry of stratospheric  $O_3$ , and the possible inadvertent depletion of  $O_3$  owing to various anthropogenic activities (e.g., high flying supersonic transports, man-made chlorofluoromethanes (CFMs), and nitrogen fertilizers used in agriculture). About 90% of the total atmospheric  $O_3$  is found in the stratosphere. As was already pointed out, stratospheric  $O_3$  protects the surface of our planet from solar ultraviolet radiation (200–300 nm). The production of stratospheric  $O_3$  is initiated by



decompose, and subsequently form new sulfate aerosols.

A number of possible mechanistic roles have been suggested for the particles in the middle atmosphere. They may (1) serve as reaction sites for gases, (2) act as a sink for gases, (3) influence the global radiation balance, (4) nucleate noctilucent clouds, and (5) provide a link between solar emission variations and tropospheric weather. Of these possibilities, only the second and third are well established at present, although the others are under active experimental and theoretical investigation.

Since the discovery of the permanent, ubiquitous stratospheric sulfate aerosol layer by Junge and coworkers two decades ago, these particles have been a subject of intense study using both *in situ* and remote sensing techniques. On the basis of such observations, we can say that the aerosols are probably composed of a 75% sulfuric acid aqueous solution, with an admixture of a variety of materials ranging from meteoric debris to nitrilos compounds derived from stratospheric  $\text{NO}_x$ . The mode of nucleation of the aerosols is uncertain and may involve tropospheric (Aitken), meteoric, ionic, and sulfur radical nuclei. The growth of the aerosols is largely controlled by the oxidation of sulfur-bearing gases which are transported into the stratosphere from the troposphere. The primary gaseous aerosol precursors are sulfur dioxide and carbonyl sulfide, both of which may have significant anthropogenic sources.

The effects of stratospheric aerosols on the global radiation balance are most noticeable following major volcanic eruptions, when the aerosol layer may be enhanced to the point that substantial temperature variations occur at the earth's surface. The May 18, 1980, eruption of Mt. St. Helens created dust-darkened skies and brilliant sunsets, dramatically demonstrating the optical perturbations that can be caused by volcanic aerosols. Recently, an increasing trend in the optical density of the background (non-volcanic) aerosol layer has been identified; one possible explanation is man's increasing usage of fossil fuel, with the attendant release of sulfur compounds to the atmosphere.

In a talk entitled 'Photochemistry of the Mesosphere and Thermosphere,' Douglas G. Torr (Utah State University) pointed out the influence of new data, largely acquired or stimulated by satellite experiments, on the understanding of thermospheric chemistry. At the end of 1979, that chemistry appeared to have crystallized into a clearly understood form, mainly as a result of analyses conducted with data taken by the Atmospheric Explorer (AE) C, D, and E satellites. To maintain consistency with laboratory measurements of the processes that destroy the  $\text{N}_2^+$  ionization, it was found that the rate coefficient for the charge exchange reaction



must be less than  $10^{-10} \text{ cm}^3 \text{ s}^{-1}$ , that is, an order of magnitude smaller than earlier laboratory measurements conducted by using nonthermal  $\text{O}^+$  ions. However, during the course of the last year new laboratory measurements established  $k$ , to be  $\sim 8 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$  (i.e., an order of magnitude larger than the aeronomically deduced value). Inclusion of this new value for  $k$ , in the ion chemistry introduced several problems, namely excess production of  $\text{N}_2^+$  ions and a shortfall in production of  $\text{O}^+(S)$  ions, which caused a deficiency in the concentrations of both  $\text{NO}^+$  and  $\text{O}_2^+$  ions. These appear now to have been simultaneously resolved with the inclusion in theoretical models of rapid charge exchange between vibrationally excited  $\text{N}_2^+$  and oxygen atoms. The previously anomalous seasonal behavior of the  $\text{F}_2$  layer on a global scale appears also to be explained by this new approach.

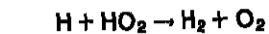
Torr reported that a major step forward in the laboratory measurement of rate coefficients was that of E. C. Ziff (University of Pittsburgh), who used laser induced photofluorescence techniques to study the behavior of  $\text{N}_2^+$  ions in specifically identified vibrational and rotational quantum states. The specific dissociative recombination coefficients for the  $v = 0, 1, 2$  vibrational levels of the  $\text{N}_2^+$  ( $X^2\Sigma^+$ ) state were found to be nearly equal in magnitude, thereby placing significant constraints on allowable thermophysical models. In the case of  $\text{O}_2^+$ , Ziff used indirect techniques to study the production of  $\text{O}^+(S)$  via  $\text{O}_2^+$  recombination and found very little production in the  $v = 0$  level, implying that high  $\text{O}^+(S)$  yields are associated with recombination in high vibrational levels. The  $\text{O}^+(S)$  yield inferred from the AE data is large, but vibrationally excited  $\text{O}_2^+$  is believed to be strongly quenched by atomic oxygen in the thermosphere. To explain this inconsistency, a dependence of the  $\text{O}^+(S)$  yield on electron temperature has been tentatively suggested.

In the area of neutral thermospheric chemistry, recent developments include the finding that the forbidden predissociation of the numerous  $\Pi_u$  and  $\Sigma_u$  valence and Rydberg states of  $\text{N}_2$  in the 11–24 eV range via radiation entrainment in an optically thick atmosphere is the dominant mechanism for  $\text{N}_2$  atom production, the finding that destruction of  $\text{N}_2^+(D)$  constitutes a major source of  $\text{O} 6300 \text{ \AA}$  dayglow, and the discovery of an oxygen geocorona of  $\sim 3000$  to  $4000 \text{ km}$ .

Mesospheric chemistry has not received as much recent emphasis as has thermospheric chemistry, mainly because the necessary experimental effort has been planned for the 1980's. Some recent results

have provided new insights, however. A reevaluation of the  $\text{O}_2$  dissociation rate has yielded ratios of 1 and 0.6 for the old to new rates at the altitudes 80 and 50 km, respectively. Similar results for the dissociation rate for water vapor indicate that the uncertainties are such that the actual rates may be as low as 0.46 or as high as 1.65 times the currently accepted values.

In an analysis of solar proton events, it has been recently pointed out that above  $\sim 75 \text{ km}$  the rate of dissociation of water vapor produced by recombination of hydrated ions may be large enough to deplete  $\text{H}_2\text{O}$  concentrations significantly, since the odd hydrogen at these altitudes is effectively lost in conversion of  $\text{H}_2$  by the reaction



The net result would be an initial depletion followed by a storm time increase in ozone.

#### Terrestrial Planets: Venus and Mars

Planetary atmospheric chemistry at the symposium was introduced by Ronald G. Prinn (Massachusetts Institute of Technology), who discussed the 'Chemistry of the Atmosphere of Venus.' The atmosphere is dominated by  $\text{CO}_2$ ,  $\text{N}_2$  at  $\sim 3.4\%$  being the next most abundant constituent.  $\text{HCl}$  and  $\text{HF}$  were detected by ground-based spectroscopy in 1980; their presence is consistent with what one would derive by heating earth rocks to the 750°K Venusian surface temperature.  $\text{CO}$  is also present, primarily as a result of  $\text{CO}_2$  photolysis; other suggested sources are lightning and thermochemistry near the surface. Water vapor is present at concentrations of a few parts per million above the clouds and a few hundred parts per million below the clouds.

The recent *Venera* and Pioneer Venus probes confirmed earlier suggestions of a rich sulfur chemistry in the Venusian atmosphere by detecting  $\text{SO}_2$  and, more tentatively,  $\text{H}_2\text{S}$ ,  $\text{S}_2$ , and  $\text{S}_8$  below the clouds. Collaborative evidence for concentrated sulfuric acid as a major component of the clouds of Venus was also provided. The ultimate source of the sulfur is undoubtedly outgassing of the crust, perhaps partially by volcanism. This outgassing is expected to be in the form of  $\text{H}_2\text{S}$  and  $\text{OCS}$ . The latter gas has not yet been observed but would be expected at altitudes below 20 km (where measurements are presently lacking) due to equilibration of the observed  $\text{CO}$ ,  $\text{S}_2$ ,  $\text{S}_8$ , and  $\text{SO}_2$ .

It appears that photochemical oxidation of  $\text{SO}_2$  and, to a lesser extent,  $\text{H}_2\text{S}$  is the major source of the sulfuric acid in the clouds and the major sink for the  $\text{O}_2$  produced from  $\text{CO}_2$  photodissociation at high altitudes. The major sink for  $\text{CO}$  appears to be oxidation to  $\text{CO}_2$  by reactions with  $\text{SO}_3$ ,  $\text{SO}_2$ , and  $\text{SO}$  near the surface. The *Venera* spectrophotometer indicated that the water vapor mixing ratio decreases as one approaches the surface. This property may be linked to photodissociation of  $\text{S}_2$  and  $\text{S}_8$  by near ultraviolet and visible light below the clouds. Collaborative laboratory studies are required.

Next to the earth's present atmosphere, that of Mars is probably the best understood. In a talk entitled 'Photochemistry of the Martian Atmosphere,' T. Y. Kong (Bell Laboratories) pointed out some of the distinctions between them. The atmosphere of the earth is controlled by four processes: photochemical, physical, biological, and anthropogenic. On Mars, only the first two of these appear to exist. As a result, Mars has not developed an oxygen-rich atmosphere, but maintains a thin atmosphere (surface pressure 8 mbar) dominated (96%) by  $\text{CO}_2$ . About 4% of the atmosphere is  $\text{N}_2$  and  $\text{Ar}$ .

The low opacity of the Martian atmosphere (except during dust storms) promotes photochemical processes, with such species as  $\text{O}_3$ ,  $\text{CO}$ , and  $\text{NO}$  being observed as a result. Much of the driving force for this chemistry comes from the photodissociation of  $\text{CO}_2$  and  $\text{H}_2\text{O}$ . Model calculations have been reasonably successful at treating such features as the wide variation (up to a factor of  $\sim 30$ ) of low-altitude ozone concentrations at different latitudes and seasons. At high altitudes, the dominant form of odd oxygen is predicted by the models to be the oxygen atoms. The dominant form of odd hydrogen near the ground is  $\text{HO}_2$ , not unlike the troposphere of the earth. Still to be investigated are the ways in which Martian dust influences the chemistry of the Martian atmosphere.

The chlorine chemistry of the Venus atmosphere, although studied for some years, is more poorly understood than that of the earth. The 1 ppm of  $\text{HCl}$  that is present will photolysis to produce both odd hydrogen and odd chlorine radicals. These latter species are expected to play some role in the oxidation of  $\text{SO}_2$  at cloud level, but the exact mechanisms are not yet known. The much lower  $\text{O}_2$  concentrations on Venus will render  $\text{Cl}$  much more abundant than  $\text{ClO}$ , in contrast to the situation in the earth's atmosphere.

The clouds of Venus possess a complex structure as a function of altitude (Figure 6) and appear to be composed of several different chemicals as well.

These points were discussed by Owen B. Toon

(NASA Ames Research Center) in his talk 'Chemistry of the Clouds of Venus.'

The upper clouds are largely sulfuric acid, but the particle size distribution is bimodal with peaks at  $\sim 0.1$  and  $1.0 \mu\text{m}$ , a distribution that pure sulfuric acid chemistry apparently cannot reproduce. This implies the presence of another constituent. Sulfur particles violate several observational constraints and cannot provide the opacity needed to explain the ultraviolet markings on Venus. A candidate compound is  $\text{Cl}_2$ .

The lower clouds are characterized by a trimodal

vertical structure of  $\text{N}_2$  and  $\text{SO}_2$  (Figure 7). The thermal structures of the Jovian planets (Figure 8) differ principally in the temperature offset owing to solar distance. Their composition is dominated by  $\text{H}_2$  and  $\text{He}$ . Jupiter and Saturn have recently been the subjects of *Voyager* flyby analyses; combined with ground-based observations, these indicate the presence of  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$ ,  $\text{NH}_3$ , and  $\text{PH}_3$  in the atmospheres of both planets. In addition,  $\text{CO}$  has been detected in the Jovian atmosphere.

The observations can be compared with theory by using the deduced thermal structure, assuming thorough atmospheric mixing to great depths, and invoking chemical reactions that link the observed species. For Jupiter, this is done by starting with a parcel containing a variety of volatile constituents at an interior reference level of  $p = 200 \text{ kbar}$ ,  $T = 2000^\circ\text{K}$ , allowing the parcel to ascend adiabatically, and calculating the levels at which the various liquid and solid phases condense out. The condensates are assumed to remain as aerosols at these levels. Dense water clouds are calculated to form at  $270^\circ\text{K}$ ,  $80 \text{ km}$  above the surface reference level, where  $p = 20 \text{ bar}$ . Near the  $200^\circ\text{K}$  ( $90 \text{ km}$ ) level,  $\text{He}$  is thought to react with  $\text{NH}_3$  to form a cloud of solid  $\text{NH}_4\text{SH}$  particles. White crystals of ammonia precipitate out at  $154^\circ\text{K}$  ( $p = 800 \text{ mbar}$ ,  $z = 120 \text{ km}$ ) to produce the visible upper cloud layer, a result supported by lines characteristic of solid ammonia in Jupiter's emission spectra.

It appears that many of the Jovian trace molecules are created high in the atmosphere (where solar photons are readily available) and are mixed down into the lower atmosphere. The color of the Great Red Spot and of some of the clouds is presumably a result of phosphorus chemistry, although compounds of sulfur have also been suggested. Analyses of the hydrocarbon chemistry are complicated by the fact that uncertainty in the photochemistry of  $\text{CH}_4$  and

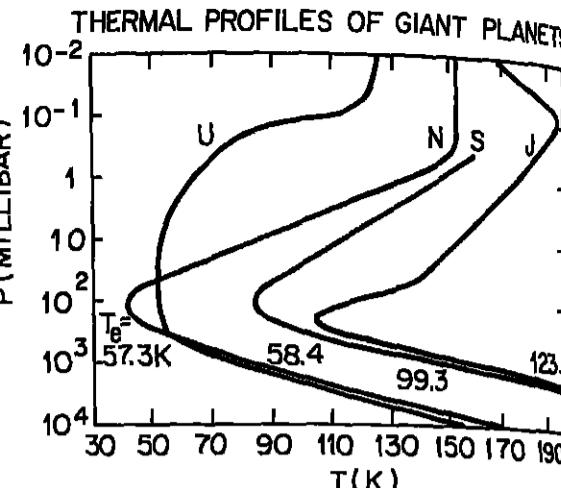


Fig. 7. The thermal structures of the Jovian planets. (D. Gautier and R. Court, *Icarus*, 59, 28, 1979; reproduced by permission of Academic Press.)

$\text{C}_2\text{H}_6$  and the possibility of chemical effects from precipitating magnetospheric particles. Ongoing analyses of the Voyager data may place constraints on some of these processes.

The photochemistry of the atmospheres of Saturn, Uranus, and Neptune have received much less attention than that of Jupiter and were not explicitly discussed at the symposium. There is every indication that similar approaches can be used to describe the chemistry of all the Jovian planets, however. All contain small hydrocarbons and probably ammonia. Spurred on by the wealth of Voyager data, they will no doubt be actively modeled over the next several years.

#### Titan, Triton, and Pluto

Little has been known of the atmospheres of these bodies, and they have received little study as a result. The paucity of information was dramatically reversed for Titan by the *Voyager* flyby in November 1980. Previously thought to contain mostly methane, the atmosphere was found to be almost entirely  $\text{N}_2$ . Trace amounts of  $\text{CH}_4$ ,  $\text{C}_2\text{H}_2$ ,  $\text{C}_2\text{H}_4$ ,  $\text{C}_2\text{H}_6$ , and  $\text{HCN}$  were also detected by *Voyager*, proving conclusively the existence of hydrocarbon photochemistry on Titan. The surface temperature and pressure, atmospheric scale height, and the presence of a  $75^\circ\text{K}$  temperature minimum at  $\sim 50 \text{ km}$  were also established. Theoretical studies aimed at matching these observations and thus establishing the atmospheric chemical mechanisms are being vigorously pursued.

Titan and Pluto are known to possess frozen methane on their surfaces. At surface temperatures of  $50^\circ$ – $70^\circ\text{K}$ , the methane vapor pressure from this methane ice will provide tenuous atmospheres. It is reasonable to suppose that the unobserved gases  $\text{N}_2$ ,  $\text{H}_2$ , and  $\text{Ar}$  will be present as well. Until more detailed experimental observations can be obtained, little information will be forthcoming on the atmospheric chemistries of Triton and Pluto.

#### Summary

The photochemistry of planetary atmospheres has come of age in the last decade. From a single example, the modern atmosphere of the earth, the vigorous programs of planetary exploration have provided data on the atmospheres of Venus, Mars, Jupiter and Io, and Saturn and Titan. In addition, studies of geological records have provided substantial inferential information on the ancient atmosphere of the earth. Although we can still look forward to the observations of *Voyager* 2 of Uranus, Neptune, and Triton, eight examples of planetary atmospheres are available to the theoretical photochemists. Their similarities and their differences provide striking examples of the concomitant diversity and scientific rigor of nature.

#### Acknowledgments

The symposium on which this report is largely based was ably chaired by R. J. McNeal, Manager of

the Air Quality Program at NASA Headquarters. We thank him for his efforts. Our thanks also go to the symposium participants, both for their careful preparation and presentation and for their comments on earlier version of this report.



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Thomas E. Graedel is a Member of Technical Staff in the Chemical Kinetics Research Department, Bell Laboratories, Murray Hill, New Jersey. He holds a B.S. (Chemical Engineering) from Kent State University, and M.S. and Ph.D. degrees (Astronomy) from the University of Michigan. His current research interests are in computer modeling of the chemistry of the earth's troposphere and in experimental and theoretical investigations of atmospheric corrosion of metals and alloys. A member of AGU since 1965, he has served as Chairman of the Geophysical Monograph Board and is presently the Chairman of the Books Board and a member of the Publications Committee.

## News

#### Volcano Organization Formed

The past decade was an unusually eventful one for volcanology, with the 1973 eruption on Heimaey, Iceland, the 1976–1977 rumblings of the Soufrière Volcano on Guadeloupe, the huffing and puffing of Iceland's Krafla Volcano in 1975, the many eruptions of Mount Etna in Sicily in the 1970's, and the reawakening of Mount St. Helens in 1980 in the United States. In addition to their scientific duties, volcanologists have had to play an important social role as advisors to administrators. For example, the political decision to evacuate Guadeloupe exposed French volcanologists to scrutiny more severe than previously experienced by members of the profession. These volcanologists met the challenge by completely reorganizing their volcano observatories and by increasing their volcano research.

In recognition of this recent reorganization and the associated modernization of volcano observatories on Guadeloupe and Martinique, the French government invited representatives from the world's volcano observatories and institutions to meet. The meeting, from February 18–21, 1981, resulted in the establishment of the World Organization of Volcano Observatories (WOVO). As its name implies, WOVO is concerned exclusively with volcano observatories and volcano monitoring; its activities will not duplicate the functions of existing international organizations, such as the International Association of Volcanology and Chemistry of the Earth's Interior (IAVCE), which focus generally on volcanology and allied topics.

The principal objectives of WOVO are to create or improve links between observatories and institutions directly involved in volcano monitoring; to facilitate the exchange of views and experiences by convening periodic, perhaps annual, meetings; to maintain an up-to-date inventory of instruments and manpower, which could be made available to any of the member institutions if a situation arises that requires scientific reinforcement; and to promote funding from international organizations, which could help defray travel and related expenses of scientific reinforcement teams.

The second working group concerned itself with defining the 'adequate minimum' equipment and personnel for a volcano observatory. Given the premise that any monitoring is better than none, the working group—composed of R. Tilling (U.S.), Chapman, M. Espenfeld (Mexico), E. Malavassi (Costa Rica), L. Villari (Italy), and J. P. Vlode (France)—unanimously agreed that continuously recording monitoring systems are absolutely essential for timely volcanic hazard assessments.

#### Luck + Merit = Grant

Granting approval to proposals submitted to the National Science Foundation (NSF) depends about as much on luck as it does on the scientific merit of the proposal, according to a study by the National Academy of Sciences' Committee on Science and Public Policy (COSPUP). In addition, concealing the names of the authors of the proposals, a practice known as 'blinding,' would not significantly change the outcome of the grant-awarding process.

The fate of a particular grant application is roughly half determined by the characteristics of the proposal and the principal investigator, and about half by apparently random elements which might be characterized as 'the luck of the reviewer draw,' according to Stephen Cole, Jonathan R. Cole, and Gary A. Simon in the COSPUP report summary, published in the November 20 *Science*. If researchers' proposals for NSF grants were rated again by an equally qualified group of reviewers, between 25% and 30% of NSF funding decisions could be reversed, they said.

However, this

low or moderate correlation between the funding decision and the prestige rank of the applicant's current academic department, academic rank, geographic location, NSF funding history during the previous 5 years, and the location of doctorate training.

Given the chance of the review process, it would appear that the more proposals a scientist submits to NSF, the more likely it is he or she will be funded. In fact, Cole, Cole, and Simon said, "eminent scientists may be more likely to be funded than less well-known ones not because their probability of success is greater for each submitted proposal, but because they submit many proposals and are not deterred by an individual rejection."

Would a "blind" review system be better? In general, COSPUP found that "it was extremely difficult to conceal the authorship of proposals," and that reviewer disagreement, not blinding, played a greater role in their study.

What does all of this mean for science? A distinction must be made between the effect of randomness in the peer review system on individual applicants and the effect on science itself, according to the report summary. While the randomness may be frustrating to individual scientists, "it may have little effect on the rate of development of science as a whole," wrote Cole, Cole and Simon. "One clear disadvantage for science of the current peer review system is that it compels even our most talented scientists to spend substantial amounts of time and energy writing proposals, time and energy that might be more fruitfully spent doing research." —BTR

#### NSF Reviewed—An Analysis

The awarding of scientific research grants jointly to investigators and their academic institutions by the National Science Foundation (NSF) (see news item above) is based on a concept of high ideals; review of proposed research plays is done by peers selected from the academic community. The peer review process and the way awards of funds are made to an investigator's institution are often mildly criticized, but among members of the community there has long been an underlying respect of the concepts, a recognition that the process tends to lead to high standards in research. As peers are chosen, a certain degree of unevenness in the review process is to be expected. Just how much unevenness has been revealed in a recent study sponsored by the National Academy of Sciences Committee on Science and Public Policy (COSPUP). The results indicate that the peer review process is random and unbiased. The results show some tendency toward unevenness between groups of peers, but the NSF process evidently is a finely tuned one, especially when compared with other more or less subjective methods of evaluation.

The choice and selection of reviewers of a proposal is, by necessity and intention, subjective in certain ways. The reviewers must know the field, the specifics of the proposed research, and they must be competent to judge the investigator and the institutional facilities. To do this, the reviewers must be subjective, bordering on a conflict of interest. And yet, the study showed there was little, if any, bias toward proposers. There was a measurable component of what the study called "luck of the reviewer draw," which was seen when a given set of research proposals was evaluated by more than one group of reviewers. If the selection of reviewers was random, then the reviews were uneven. The modifying subjectivity is the thoughtful choice of reviewers by the NSF program directions. This has been called "informed subjectivity" (*Chemical and Engineering News*, Nov. 16, 1981). Reviewer disagreement seemed to be mostly a result of real and legitimate differences of opinion among experts about what good science should be. There is no guarantee that a given proposal will be judged the same by several sets of reviewers, and yet the report states . . . this should not be interpreted as meaning either that the entire process is random or that each individual reviewer is evaluating the proposal in a random way."

Even though the NSF-peer review system has proven to be a fair judge of research proposals, the investigator whose budget is severely cut or whose proposal is turned down will get little solace from the results of a survey. That there is some degree of randomness or chance in the outcome serves to provide a smoothing function to the overall process. Chances improve with statistics and with the number of reviews of a given proposal. The report states . . .

Keyworth also has proposed the establishment of a new

given the importance of chance in the current process, clearly the more proposals a researcher submits the higher the probability of being funded. Thus there is good news that an investigator should not be deterred by the rejection of proposals.

Perhaps the most important outcome of the study is that the review process is a good one and is working. The questions or problems that arise are more sophisticated and subtle than factors revealed by a statistical study. They have to do with the correspondence between the research proposed and the research actually done, the attempt to judge creativity before it is created and the obvious question of trying to preclude a result that cannot be predicted beforehand. The review process is influenced by the "zero-based" budget syndrome; a new budget per year means a new proposal to be reviewed per year, which must not only reflect new findings per year but a new project per year. The result may be that science is being done on a short project one-per-year basis. Longer, perhaps more fundamental, and thus more vaguely proposed research projects do not fit into the system too well. NSF-funded projects of 5 years or more duration are almost unheard of today.

It is clear, however, the shorter time of a project benefits from more rapid publication of the results, more critical response by the academic community, and a sharper competitive edge. The competitive spirit today in research is such that if an investigator were to take 5 years to publish results of his research, often someone else would have been first. The high level of excellence of scientific research in the United States has resulted in part from the competitive process that, in turn, is fostered by the NSF peer review process.—PMB

#### NSF Reviewed—An Analysis

Five AGU members were granted senior Fulbright awards for university teaching and advanced research abroad for 1981–1982, according to the U.S. International Communications Agency and the Board of Foreign Scholars.

Yvonne Herman-Rosenberg, associate professor of geology at the Washington State University in Pullman, will research Black Sea Quaternary benthic foraminifera as indicators of sea-level fluctuations. The research will be conducted at the University of Bucharest in Romania from May through July, 1982.

George V. Keller, professor of geophysics and head of department at the Colorado School of Mines, is lecturing at the Moscow State University on exploration for geothermal energy. His 3-month stay in Moscow concludes this month.

Robert D. Lawrence, associate professor of geology at the Oregon State University in Corvallis, will lecture for the entire academic year at the University of Peshawar in Pakistan on structural geology.

Walter H. Munk, professor of oceanography at the Scripps Institution of Oceanography, will research ocean mapping by remote acoustic sensing for the entire academic year at the University of Cambridge in the United Kingdom.

Carl Wunsch, an oceanographer at the Massachusetts Institute of Technology, also will be at the University of Cambridge for the entire academic year. He will research the application of inverse methods, acoustic tomography, and satellite altimetry to the problem of determining ocean circulation through global measurement systems.

#### New Planet Missions May Be Halted

A December 2 report in the *Washington Post* states that George Keyworth, science advisor to President Reagan, has "recommended halting all new planetary space missions for at least the next decade—an idea he said the White House seems to be buying."

A spokesman in the Office of Science and Technology Policy (OSTP) told *Eos* that in keeping with this, astrophysics and astronomy, not planetary missions, would be emphasized in the fiscal 1983 budget. The OSTP spokesman was unsure what Keyworth meant by "an idea he said the White House seems to be buying." There has been other talk that budget cuts would jeopardize space exploration (*Eos*, October 20, p. 705).

Keyworth also has proposed the establishment of a new

#### The June Bacon-Bercey Scholarship in Atmospheric Sciences for Women 1982–1983

Expressly for women intending to make a career in the atmospheric sciences. This monetary assistance provided through a gift from June Bacon-Bercey, a noted meteorologist, will be given to a woman who shows academic achievement and promise. To qualify candidates must be one of the following:

- a first-year graduate student in an advanced degree program in atmospheric sciences;
- an undergraduate in a bachelor's degree program in atmospheric sciences who has been accepted for graduate study;
- a student at a 2-year institution offering at least 6 semester hours of atmospheric sciences, who has been accepted for a bachelor's degree program and who has completed all of the courses in atmospheric science offered at the 2-year institution.

For application forms contact:  
**American Geophysical Union**  
Member Programs Division  
2000 Florida Ave., N.W.  
Washington, D.C. 20009

462-6903  
800-424-2488 outside the Washington, D.C. area

Awardee selection will be made by the AGU Subcommittee on Women in Geophysics, in consultation with the AGU Meteorology Section.

**Application Deadline, May 1, 1982**

science advisory board at the White House. President Reagan has received the proposal, which calls for a 15-member panel similar to the President's Science Advisory Committee that was abolished by President Nixon in 1973. Reagan had not approved the proposal at deadline.—BTR

#### Geophysicists

Robert L. Bates is the 1981 recipient of the Association of Earth Science Editors Award for Outstanding Editorial or Publishing Contributions. Bates, formerly a professor of geology at The Ohio State University, is well known for *The Geologic Column*, a regular feature in *Geotimes*. Previous recipients of the award are Brian J. Skinner, Philip H. Abelson, Marie Slogar, and Edwin B. Eckel.

Philip S. Justus was promoted to section leader of the high-level radioactive waste repository section of the U.S. Nuclear Regulatory Commission in Silver Spring, Md. Before joining the Division of Waste Management, he was with the Geosciences Branch siting nuclear reactors.

John Rodgers, Silliman Professor of Geology at Yale University, was awarded the Geological Society of America's Penrose Medal for 1981. He was cited for his "innovative ideas concerning Appalachian geology." Rodgers is an AGU Fellow.

George H. Sutton, formerly professor of geophysics and associate director of the Hawaii Institute of Geophysics at the University of Hawaii, has been appointed vice president of Rondout Associates, Inc., in Stone Ridge, N.Y.

Donald L. Turcotte, president-elect of AGU's Tectonophysics Section, has been awarded the 1981 Arthur L. Day Medal of the Geological Society of America. The professor and chairman of the Cornell University Department of Geological Sciences was cited for "distinguished application of chemistry and physics to geology."

Richard M. Pearl, 68, died recently. A member of the Volcanology, Geochemistry, and Petrology Section, he belonged to AGU from 1953 to 1959 and from 1973 until his death.

Students of geomorphology are often handicapped by their unfamiliarity with the concepts of mechanics or hydraulics, on which are based much of our understanding of the processes that explain the nature of landforms. The second and third chapters of the book attempt to tackle this problem by discussing the concepts of energy, force, resistance, and the nature of fluid motion. This is not an easy task, as a large amount of material has to be covered within 50 pages: from basic definition of mass, velocity, force, etc., to complicated and empiric concepts of open channel flow. The results, unfortunately, are uneven and vary from succinct and rigorous presentation to brief, confusing, and occasionally simplistic summaries. For example, the utility, at the advanced undergraduate level, of a discussion on geomagnetism that takes only four and a half lines is, at best, debatable. On the other hand the boundary layer concept as presented in chapter 3 comes through relatively clearly, albeit in a somewhat generalized format.

Following these two prerequisite chapters comes the traditional arrangement of geomorphic processes, presented with a new insight and often unconventional format. The next two chapters on weathering and mass movements in

clude an overwhelming volume of information, frequently collected from various case studies, often summarized in tables and highlighted by diagrams. These two, and a subsequent chapter on fluvial processes, are by far the longest ones in the book. The discussion on fluvial processes concentrates on river basin hydrology, the dynamics of sediment transport, and erosion and deposition both in stream channels and on hill slopes. The presentation, even of such basic topics, is usually lucid, but the impression one gets is that of unbalanced coverage. Even the author seems to have noticed this when on page 258 he attributes the lack of a proper discussion of floodplain development to a shortage of space. Presumably the shortage of space has led to occasional tenses in presentation, which may easily confuse a reader. For example, the discussion of the sequence of channel bedform in noncohesive material, is usually lucid, but the impression one gets is that of unbalanced coverage. 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Although this book's central theme is the diagrammatic representation of soil mineral solubilities, the classical use of solubility as the sum of all species is never represented. For example, the well-known parabolic pH-dependent curve for gibbsite solubility never appears. Instead, gibbsite solubility is shown as a linear pH function in terms of free aluminum ion only. This simplified approach is consistently used throughout the book.

The old and useless concept of  $rH$  (circa 1923) has been born again in a single-term expression called  $pe + pH$  by Lindsay to define the redox state of soil systems. This expression assumes that only one hydrogen ion is stoichiometrically associated with one electron per redox reaction, and any additional hydrogens are independent acid-base reactions. There is no demonstrable basis for this arbitrary rearrangement of the Nernst equation, and it leads to the confusing situation of treating pH as both a constant and a variable in the same reaction. This occurs in the chapter on sulfur, for example. It is unfortunate that  $pe + pH$  has been

developed as a main feature of the book, since it is neither new nor useful.

The solubility, redox, and distribution diagrams are all based on a very extensive tabulation of thermodynamic data compiled by M. Sadiq and W. Lindsay. This carefully compiled reference is a valuable asset of the book except for several errors occurring in it, such as (1) some free-energy values for neutral aqueous species turn out to be the values for the fully dissociated solute rather than the ion pair, triplet, etc.; (2) goethite is taken to be more stable than hematite, contrary to both field observation and reliable solubility studies; (3) several inorganic compounds that are not known to occur as minerals and have no known importance in soil chemistry are included in the solubility diagrams; and (4) the thermodynamic values are not critically selected or fully evaluated for several systems of interest to this reviewer. Nevertheless, the thermodynamic data are up to date (as of April 1978), extensive, and fully documented.

D. Kirk Nordstrom is with the Water Resources Division of the U.S. Geological Survey in Menlo Park, Calif.

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### POSITIONS AVAILABLE

**University of Utah: Faculty Positions.** The Department of Geology and Geophysics invites applications for four tenure track positions at the assistant to associate professor level.

1) **Economic Geology.** The specific area of expertise is open, however, preference will be given to candidates whose research interests are in geological, geochemical, or petrological characteristics of metallic mineral deposit.

2) **Sedimentary Geology.** Applicants should have research interests in modern or ancient sedimentary basins.

3) **Seismology.** Applicants with backgrounds and specialties in seismic reflection, seismic imaging or theoretical seismology will be given preference.

4) **Potential fields.** Geophysicist with specialty in potential theory including gravity and magnetism. (The closing date for the position is January 31, 1982).

A Ph.D. or equivalent is required. The vacancies are to be filled by September 1982; the closing date for applications for positions 1-3 is December 31, 1981. Applicants should submit a vita, transcripts, a letter describing their research teaching goals, and names of five persons for reference to William P. Nash, Chairman, Department of Geology and Geophysics, University of Utah, Salt Lake City, Utah 84112.

The University of Utah is an equal opportunity/affirmative action employer.

**Faculty Positions.** Two Faculty Positions in Geology. Tenure-track positions in geology, assistant professors. Ph.D. preferred or equivalent experience. Fall 1982.

**Petrologist/Mineralogist.** Candidate must be able to teach introductory geology, mineralogy, petrology, geochemistry, and optical mineralogy/petrography.

**Invertebrate Palaeontologist/Soft-Rock Geologist.** Candidate must be able to teach courses in invertebrate paleontology, microfossils, sedimentation, and historical geology. Additional expertise in recent marine environments highly desirable.

Applicants are expected to do research in their areas of expertise, and to lead students' field trips. Strong teaching and research commitments expected. Submit applications with resume and copies of transcripts, and have three letters of recommendation sent to the Chairperson, Department of Earth & Space Sciences, Indiana University-Purdue University at Fort Wayne, Fort Wayne, Indiana 46805. Indiana University-Purdue University is an equal opportunity/affirmative action employer.

**Research Associate/Theoretical Physical Oceanography.** Applications are invited for two postdoctoral research associate positions in the School of Oceanography, Oregon State University. Applicants will conduct research in theoretical modeling and observational comparisons of ocean circulation. Ph.D. in mathematics or the physical sciences. Submit resume, brief statement of research interests and three letters of recommendation to: Dr. Peter P. Niiler, School of Oceanography, Oregon State University, Corvallis, Oregon 97331. Closing date extended to 30 January 1982.

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**Positions in Oceanography/VIMS.** The Virginia Institute of Marine Sciences (VIMS) School of Marine Science invites applications for two state funded, oceanography research and teaching positions at the levels of Senior Marine Scientist. VIMS is a broad-based marine sciences establishment with a mission to provide sound and timely advice to executive agencies and the legislature and to conduct science research programs. The School of Marine Science offers M.A. and Ph.D. programs with a faculty of 65 and 139 graduate students.

**HEAD, DEPARTMENT OF GEOLOGICAL OCEANOGRAPHY (#119)**

Applicants are sought with research interests in estuarine sedimentary geochemistry, dynamics of cohesive sediment transport, or estuarine and coastal morphodynamics. For further information contact Dr. Robert Byrum (VIMS), 804-842-2111 (Ext. 173).

**ESTUARINE AND COASTAL HYDRODYNAMICS (Position #204)**

A physical oceanographer with a strong interest in interdisciplinary approaches to complex estuarine and continental shelf problems is desired. For further information contact Dr. Bruce Neelan (VIMS), 804-842-6131 (Ext. 244).

Candidates for both positions should have established research credentials and be dedicated to furthering the research and educational programs of the Institute. Demonstrated ability to generate extramural support is expected. Salary range is \$24,972 to \$34,107 and faculty rank is commensurate with qualifications.

Applicants should send a comprehensive curriculum vitae, reprints, and at least three letters of recommendation by February 1, 1982, stating specific position of interest, to: Employment Manager, Personnel Office, College of Wm. & Mary, Williamsburg, VA 23185.

An equal opportunity/affirmative action employer.

**Physical Oceanographer.** Royal Roads Military College expects to have a tenure track vacancy in Department of Physics effective 1 July 1982. Candidates should hold doctorate or near doctorate in physical oceanography preferably with experience in digital hardware and microcomputer applications. Appointment expected to be made at assistant professor level but salary and rank dependent on qualifications and experience. Relocation expenses can be provided. Duties include undergraduate teaching in physics and physical oceanography, and research in marine science. Applications should include complete dossier and names of three references and be sent to: Dr. E. S. Graham, Principal, Royal Roads Military College, FMD Victoria, B.C. V0S 1B0.

This competition is open to both men and women. Knowledge of English only is required. Only Canadian citizens or Landed Immigrants need apply. toute information relative à ce concours est disponible en français et peut être obtenue en écrivant à Dr. Graham.

A Ph.D. or equivalent is required. The vacancies are to be filled by September 1982; the closing date for applications for positions 1-3 is December 31, 1981. Applicants should submit a vita, transcripts, a letter describing their research teaching goals, and names of five persons for reference to William P. Nash, Chairman, Department of Geology and Geophysics, University of Utah, Salt Lake City, Utah 84112.

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Applicants are expected to do research in their areas of expertise, and to lead students' field trips. Strong teaching and research commitments expected. Submit applications with resume and copies of transcripts, and have three letters of recommendation sent to the Chairperson, Department of Earth & Space Sciences, Indiana University-Purdue University at Fort Wayne, Fort Wayne, Indiana 46805. Indiana University-Purdue University is an equal opportunity/affirmative action employer.

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**WOODS HOLE OCEANOGRAPHIC INSTITUTION**

**Woods Hole, MA 02548**  
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The strength of this book is the straightforward derivation of solubility diagrams from thermodynamic data, and as such it would be useful as a teaching reference. The weakness of the book is the lack of real world applications and relevance. Students, for example, might be led to believe that minerals such as fayalite, wustite, forsterite, and/or enstatite might be realistic solubility controls on soil water chemistry. Also the full complexity of heterogeneous soil systems with multiple simultaneous equilibria is not presented, and the effects of surface chemistry are absent. As a course textbook it has many shortcomings; but as a reference for teachers and researchers who are investigating mineral solubilities, it can be useful provided that the data base and the solubility diagrams are not accepted uncritically.

**Research Associate/Theoretical Physical Oceanography.** Applications invited for two postdoctoral research associate positions in the School of Oceanography, Oregon State University. Applicant will conduct research in theoretical modeling and observational comparisons of ocean circulation. Ph.D. in mathematics or the physical sciences. Submit resume, brief statement of research interests and three letters of recommendation to: Dr. Peter P. Niiler, School of Oceanography, Oregon State University, Corvallis, Oregon 97331. Closing date extended to 30 January 1982.

**Research Associate/Electron Microprobe.** The Electron Microprobe Center, Texas A&M University invites application for the position of Electron Microprobe Specialist. Applicants should possess a working knowledge of WDS and EDS spectrometers and accompanying computer and software programs and preferably have had experience in the Geological Sciences.

The primary duties of the position are to oversee

and maintain (with the aid of service contracts) the electron microprobe and ancillary equipment and to assist in teaching graduate courses in electron microscopy, and software programs and preferably have had experience in the Geological Sciences.

The University of Oklahoma has made a major commitment to diversify the program in the School of Geology & Geophysics. As a result five tenure-track positions are open for the fall of 1982. Six new faculty were added to the School in the fall of 1981 (bringing the total full-time faculty to 15), and an additional three positions will be available during 1982-1983. A new building that will house the School is in the design stage, and the successful applicant will be involved in the planning and construction of the new building.

An affirmative action/equal opportunity employer.

**Faculty Position/ASU.** The Department of Chemistry at Arizona State University invites applications for a possible tenure track position at the assistant professor level in one of the following areas: (1) Synthetic Solid State Chemistry; (2) Surface Chemistry; and (3) Atmospheric or Low-Temperature Geochemistry. Candidates should have demonstrated in their Ph.D. and/or postdoctoral work the ability to develop a vigorous and innovative research program in one or more of the above areas and have a commitment to instructional excellence. A resume, brief description of research plans, and three letters of recommendation should be sent to Professor William S. Glaeser, Chairman, Search Committee, Department of Chemistry, Arizona State University, Tempe, Arizona 85287. EDAA employer.

**Structural Geologist/University of Illinois at Urbana-Champaign.** (Search received) The Geology Department is seeking a structural geologist for a tenure-track (assistant professor) faculty position. A Ph.D. is required. Salary open. The successful candidate will be expected to teach advanced undergraduate and graduate courses in structural geology and establish a research program. For equal consideration, applications, including the names of three referees, should be sent by February 1, 1982 to Dr. D. E. Anderson, Department of Geology, University of Illinois, 245 Natural History Building, 1301 West Green Street, Urbana, IL 61801-2998, (217) 333-6713.

Position to be filled by September 1982.

The University of Illinois is an affirmative action/equal opportunity employer.

**Structural Geologist/University of Wyoming.** (Search received) The Geology Department seeks applicants for a tenure track appointment in structural geology expected to be available beginning fall semester 1982 or earlier. Duties will include teaching of undergraduate and graduate courses in structural geology, supervising M.S. and PhD theses, and research in structural geology. An EDD degree is required. Salary open. The successful candidate will be expected to teach advanced undergraduate and graduate courses in structural geology and establish a research program. For equal consideration, applications, including the names of three referees, should be sent by February 1, 1982 to Dr. D. E. Anderson, Department of Geology, University of Illinois, 245 Natural History Building, 1301 West Green Street, Urbana, IL 61801-2998, (217) 333-6713.

Position to be filled by September 1982.

The University of Wyoming does not discriminate on the basis of race, sex, or sex, and is an equal opportunity employer.

**Seismologist/University of Utah.** Search extended; the University of Utah is expanding its geophysics program in the Department of Geology and Geophysics by adding a tenure-track faculty member in seismology at the assistant professor level. Applicants with strong backgrounds and specialties in seismological theory and/or seismic wave propagation are invited to apply. The successful candidate will be expected to teach undergraduate and graduate courses, and to pursue an active research program with graduate students. The department has modern teaching and research programs in geology and geophysics, and has close associations with the numerical analysis and data processing groups in computer science, electrical engineering and mathematics. The geophysics component of the department has strong research and teaching programs in seismology, electrical and electromagnetic methods, thermal properties of the earth, and potential fields. Current research in seismology includes seismological and earthquake research utilizing a new P-wave 1170 computer with plotter and terminal; monitoring of the intermediate seismic belt by a 55 station seismometer network utilizing a new on-line PDP 11/34 computer; major expansions in seismic reflection profiling investigations of sedimentary basins from seismic tomography; application of inverse theory to seismic wave properties of volcanic systems and seismic research in tectonophysics. The closing date for applications is December 31, 1981. A Ph.D. is required for this position. Applicants should submit a vita, transcripts, a letter of reference, and three letters of recommendation to: Dr. Robert S. Houston/Head, Department of Geology and Geophysics, University of Wyoming, Laramie, Wyoming 82071-3006.

The University of Wyoming is an equal opportunity/affirmative action employer.

**Faculty Position in Geophysics/Structural Geology/Engineering Geology.** The Department of Geological Sciences at Case Western Reserve University in Cleveland, Ohio is seeking candidates to fill an anticipated faculty position in the broadly defined areas of geophysics/structural geology/engineering geology. While field of specialization is open, the successful candidate will be charged with conducting research in geophysics and engineering geology, and to pursue an active research program with graduate students. The individual will be expected to teach undergraduate and graduate courses, and to pursue an active research program with graduate students. The department has modern teaching and research programs in geology and geophysics, and has close associations with the numerical analysis and data processing groups in computer science, electrical engineering and mathematics. The geophysics component of the department has strong research and teaching programs in seismology, electrical and electromagnetic methods, thermal properties of the earth, and potential fields. Current research in seismology includes seismological and earthquake research utilizing a new P-wave 1170 computer with plotter and terminal; monitoring of the intermediate seismic belt by a 55 station seismometer network utilizing a new on-line PDP 11/34 computer; major expansions in seismic reflection profiling investigations of sedimentary basins from seismic tomography; application of inverse theory to seismic wave properties of volcanic systems and seismic research in tectonophysics. The closing date for applications is December 31, 1981. A Ph.D. is required for this position. Applicants should submit a vita, transcripts, a letter of reference, and three letters of recommendation to: Dr. Robert S. Houston/Head, Department of Geological Sciences, Case Western Reserve University, Cleveland, Ohio 44106.

Case Western Reserve University is an equal opportunity/affirmative action employer.

**Graduate Research Assistantships in Physical Oceanography.** Opportunities for graduate study with research assistantships available for students interested in M.S. or Ph.D. programs. An summer program with stipend is open to college juniors. Write Douglas Caldwell, School of Oceanography, Oregon State University, Corvallis OR 97331.

**Exxon Teaching Fellowship at University of Michigan Geological Sciences.** Applications are invited for a three-year fellowship in the Exxon Education Foundation, supported by the Exxon Education Foundation. Annual stipends will be \$12,000, \$13,500, and \$15,000 for the first, second, and third years, respectively, with full waivers for tuition and fees. The successful applicant will be a person with the ability to teach geology at the graduate and undergraduate levels, in addition to carrying out a vigorous research program. Ample opportunities exist for research collaboration both within the Department of Geological Sciences and with faculty members in the School of Engineering.

Ph.D. or equivalent is required. Please submit applications, consisting of résumé, names of three references and a statement of research and teaching interests to:

Samuel M. Savin  
Department of Geological Sciences  
Case Western Reserve University  
Cleveland, Ohio 44106.

Case Western Reserve University is an equal opportunity/affirmative action employer.

**Seismologist/University of Utah.** Search extended; the University of Utah is expanding its geophysics program in the Department of Geology and Geophysics by adding a tenure-track faculty member in seismology at the assistant professor level. Applicants with strong backgrounds and specialties in seismological theory and/or seismic wave propagation are invited to apply. The successful candidate will be expected to teach undergraduate and graduate courses, and to pursue an active research program with graduate students. The department has modern teaching and research programs in geology and geophysics, and has close associations with the numerical analysis and data processing groups in computer science, electrical engineering and mathematics. The geophysics component of the department has strong research and teaching programs in seismology, electrical and electromagnetic methods, thermal properties of the earth, and potential fields. Current research in seismology includes seismological and earthquake research utilizing a new P-wave 1170 computer with plotter and terminal; monitoring of the intermediate seismic belt by a 55 station seismometer network utilizing a new on-line PDP 11/34 computer; major expansions in seismic reflection profiling investigations of sedimentary basins from seismic tomography; application of inverse theory to seismic wave properties of volcanic systems and seismic research in tectonophysics. The closing date for applications is December 31, 1981. A Ph.D. is required for this position. Applicants should submit a vita, transcripts, a letter of reference, and three letters of recommendation to: Dr. Robert S. Houston/Head, Department of Geological Sciences, Case Western Reserve University, Cleveland, Ohio 44106.

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**Graduate Study in Oceanography Oceanside Engineering.** Research Assistantships and research fellowships are available for graduate study in Physical and Chemical Oceanography, Oceanographic Engineering, and Marine Geology and Geophysics leading to a Ph.D. or Sc.D. degree, available to students at Woods Hole Oceanographic Institution and the Massachusetts Institute of Technology. The awards cover tuition and provide an average monthly (tuition) stipend of \$540 to \$590. Research topics available to students reflect the interests of the more than 100 doctoral scientists and engineers at WHOI and the faculties of ten different departments at MIT.

The program encourages applications from students with an undergraduate degree





waves and no clear circulation has been found between tides produced by the planet and the tides produced by the Sun. The IAP-1 and -2 IAPWDS places anomalies along with wave profiles obtained by the IAP-1 and -2 tides.

In addition measurements made by the project were used to determine the intrinsic properties of these waves, using the spectrum and two dipole and a uniform, anti-symmetrical magnetic field. The model's spectrum for the convection electric field with equal-potential magnetic field lines and with an asymptotic behavior such that at a great distance the electric field is zero. The convection electric field is directed along the x-axis toward the south. This is in direct with the one predicted from the model. The model's spectrum for the magnetic field is zero. The dipole shifting effect, positive, demonstrate that these are right-hand elliptically polarized whistlers. The model's spectrum for the frequency and wavelengths  $\lambda$  is  $10^3$  cm. The wave length propagates in the plane formed by the x and y direction and the solar wind angle. The model's spectrum for the magnetic field is zero. The wavelengths estimated from cold plasma theory for waves of these frequencies traveling in the observed directions are often not sufficient to account for the observed amplitudes at the times of observation. This argues against the possibility that the waves could have propagated upstream to the upstream position from the two shock structures, upstream.

J. Geophys. Res., Blue, Paper 1A1722

#### 3720 Interactions between solar wind and magnetosphere

W. C. SWEETMAN AND J. HANCOCK, INSTITUTE FOR GEOPHYSICS AND ASTRONOMY, UNIVERSITY OF TORONTO, TORONTO, ONTARIO, N3B 2P8, CANADA

E. C. WHITBRETT (Center for Astrophysics and Space Sciences, University of California at San Diego, La Jolla, California 92093, and

H. B. SKELTON

We calculate the current contributed to the magnetosphere by the solar wind in the vicinity of a magnetic storm. The current is found to be

10% of the current in the magnetosphere at

the Alfvénic zone.

J. Geophys. Res., Blue, Paper 1A1723

We present preliminary results of applying the Rice Convection Model to the early-morn convection electric field and the magnetic storm of July 20, 1977. The model computes electric fields and currents, as well as plasma distributions and velocities, in the ionosphere-magnetosphere system. In the equatorial region the electric field is directed along the x-axis toward the south. The convection electric field is directed along the x-axis toward the south. The convection electric field is zero. The dipole shifting effect, positive, demonstrate that these are right-hand elliptically polarized whistlers. The model's spectrum for the frequency and wavelengths  $\lambda$  is  $10^3$  cm. The wave length propagates in the plane formed by the x and y direction and the solar wind angle. The model's spectrum for the magnetic field is zero. The wavelengths estimated from cold plasma theory for waves of these frequencies traveling in the observed directions are often not sufficient to account for the observed amplitudes at the times of observation. This argues against the possibility that the waves could have propagated upstream to the upstream position from the two shock structures, upstream.

J. Geophys. Res., Blue, Paper 1A1723

#### 3733 Magnetic storms

COMPUTER SIMULATION OF INNER MAGNETOSPHERIC DYNAMICS FOR THE MAGNETIC STORM OF JULY 20, 1977

R. A. WOLF (Space Physics and Astronomy Dept., Rice University, Houston, Texas 77003), R. H. HUMPHREY, G. M. VIEHL, G. M. HALLIF, AND

C. E. COHEN

We calculate the current contributed to the magnetosphere by the solar wind in the vicinity of a magnetic storm. The current is found to be

10% of the current in the magnetosphere at

the Alfvénic zone.

J. Geophys. Res., Blue, Paper 1A1723

#### EXPLANATIONS

This Calendar continues the series begun for the IGY years 1957-58, and is issued annually to recommend data for solar and geophysical observations which cannot be carried out continuously. Thus, the amount of observational data in existence tends to be larger on Calendar days. The case of vertical incidence sounding, the need to obtain quarterly ionograms at many stations as possible is particularly discussed, and likely prior recommendation (a) below when not for vertical incidence.

For the vertical incidence (VI) sounding program, the summary recommendations are: (a) all stations should make soundings at least every quarter hour. Sounding which normally record at every quarter should, if possible, record more frequently on RWDS; (b) all stations are encouraged to make i-plots on RWDS; (c) plots should be made for high-latitude stations, and for the selected "representative" stations at lower latitudes for days (a), including RWDS and WGDs; (d) continuous real-time data are acceptable in place of i-plots at appropriate times for QWDSs to be sent to WGDs; (e) stations in the eclipse zone and its conjugate area should take continuous observations on solar eclipses days and special observations on adjacent days. See also recommendations under Airglow and Aurora Phenomena.

For incoherent scatter observation program, every effort should be made to obtain measurements at least on the Incoherent Scatter Coordinated Observation Days, and intensive series should be attempted whenever possible in WGDs or the Airglow and Aurora Periods. The need for collateral VI observations with not more than quarter-hourly spacing at least during all observation periods is stressed. Dr. M. A. Baron (U.S.A.), USRI Working Group G.5, is coordinating special programs.

For the ionospheric drift or wind measurement by the various radio techniques, observations are recommended to be concentrated on the weeks including RWDS.

For traveling ionosphere disturbances propose special periods for coordinated measurements of gravity waves induced by magnetospheric activity, probably on selected RWDS and WGD.

For the ionospheric absorption program half-hourly observations are made at least on all RWDS and half-hourly i-plots sent to WGDs. Observations should be continuous on solar eclipse days for stations in eclipse zone and in its conjugate area. Special efforts should be made to obtain daily absorption measurements at temperate latitude stations during the period of absorption Winter Anomaly, particularly on days of abnormally high or abnormally low absorption (approximately October-March, Northern Hemisphere; April-September, Southern Hemisphere).

For back-scatter and forward-scatter programs, observations should be made and analyzed on all RWDS at least.

For synoptic observations of mesospheric (D region) electron densities, several groups have agreed on using the RHD for the month around noon.

For ELF noise measurements involving the earth-ionosphere cavity resonances any special effort should be concentrated during the WGDs.

It is recommended that more intensive observations in all programs be concentrated on days of unusual meter activity.

Meteorology. Particular efforts should be made to carry out an intensified program on the RHD — such Wednesday, UT. A desirable goal would be the scheduling of meteorological rocketsondes, ozone sondes and radiometer sondes on these days, together with maximum-altitude rawinsonde ascents at both 0000 and 1200 UT.

During WGD and STRATWARM Alert Intervals, intensified programs are also desirable, preferably by the implementation of RGD-type programs (see above) on Mondays and Fridays, as well as on Wednesdays.

Middle Atmosphere Program (MAP). MAP runs from 1 January 1982 through 1985. Requirements for observing the middle atmosphere should concentrate on their observations on the RGDs, PRWDs, and QWDSs. It is recommended that observing runs for studies of planetary waves and tides be at least 10 days centered on the PRWDs and QWDSs. Non-continuous studies of stratospheric warmings and the effects of geomagnetic activity on the middle atmosphere must be initiated by STRATWARM and MAGSTORM state, respectively. For more details see the "Recommended Scientific Program" on the reverse of the Middle Atmosphere Dynamics Calendar for 1982, which will be published as a special issue of the IGD for 1982.

Solar Phenomena. Observations making specialized studies of solar phenomena, particularly using new or complex techniques, such that continuous observation or reporting is impractical, are requested to make special efforts to provide to WGDs data for solar eclipse days, RWDSs and DONGTON/FLARE ALERTS. The attention of those recording solar noise spectra, solar magnetic fields and doing specialized optical studies is particularly drawn to this recommendation.

Space Research, Interplanetary Phenomena, Cosmic Rays, Aeronomy. Experimenters should take into account the observational effort in other disciplines tends to be intensified on the days marked on the Calendar, and schedule balloon and rocket experiments accordingly. If there are no other geophysical reasons for choice, in particular, it is desirable to make rocket flights as possible; where feasible, experimenters should endeavor to launch rockets to at least normal conditions on the Quarterly World Days (QWD) or RWDS, since these are also days when there will be maximum support from ground observations. Also, special efforts should be made to assure recording of telemetry on QWD and Airglow and Aurora Periods of experiments on satellites and on spaceflights in orbit around the Sun.

Ionospheric Phenomena. Special attention is continuing to particle events which cannot be forecast in advance with reasonable certainty. These will be

continued on the RGDs.

Additional copies are available upon request to IUDWDS Chairman, Dr. P. Simon, Observatoire de Paris, 6756 Meudon, France, or IUDWDS Acting Secretary, G. Roper (USA) for the 1982 calendar.

The International Uragram and World Days Service (IUDWDS) is a permanent scientific service of the International Union of Radio Science (IURS), with the participation of the International Astronomical Union and the International Union Geodesy and Geophysics. IUDWDS coordinates the International aspects of the world days program and rapid data interchange.

This Calendar for 1982 has been drawn up by J. V. Lincoln, of the IUDWDS Steering Committee, in close association with A. H. Shapley, Chairman of MONSEE of IGY, 1957-58, and has been published in various widely available scientific publications.

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#### 5735 Plasma instabilities

INFLUENCE OF ELECTRON BEAM INSTABILITY GROWTH RATES ON THE ION-PLASMA SYSTEM PARAMETERS

J. M. STRADIN (Cooperative Institute for Research in Environmental Sciences, University of Colorado/NOAA, Boulder, CO 80303)

Electron beam instabilities are studied using a simple model for an electron beam streaming through a plasma with a constant density and perpendicular to the ambient magnetic field.

Representative fluctuation spectra are computed for daytime asynchronous orbit using typical hot and cold electron beam parameters. Dips in antenna lengths of 50 m and 100 m. It is suggested that the formalism developed here may be applicable to weak electron cyclotron emission at higher latitudes, and to the generation of low-frequency ion cyclotron fluctuations by modifying the theory to ion concentrations, thermal fluctuations, magnetosphere.

J. Geophys. Res., Blue, Paper 1A1724

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COUPLED SIMULATION OF INNER MAGNETOSPHERIC DYNAMICS FOR THE MAGNETIC STORM OF JULY 20, 1977

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J. Geophys. Res., Blue, Paper 1A1725

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